

**dlyte**

THE NEW CONCEPT OF POLISHING

Corrosion tests

## VISUAL EVALUATION RESULTS.

Shinnier aspect than parts treated with traditional electropolishing

## CORROSION TESTS RESULTS.

The electrochemical behaviour of samples has been studied in a highly corrosive solution ( $[\text{NaCl}] = 30 \text{ g/L}$ ).

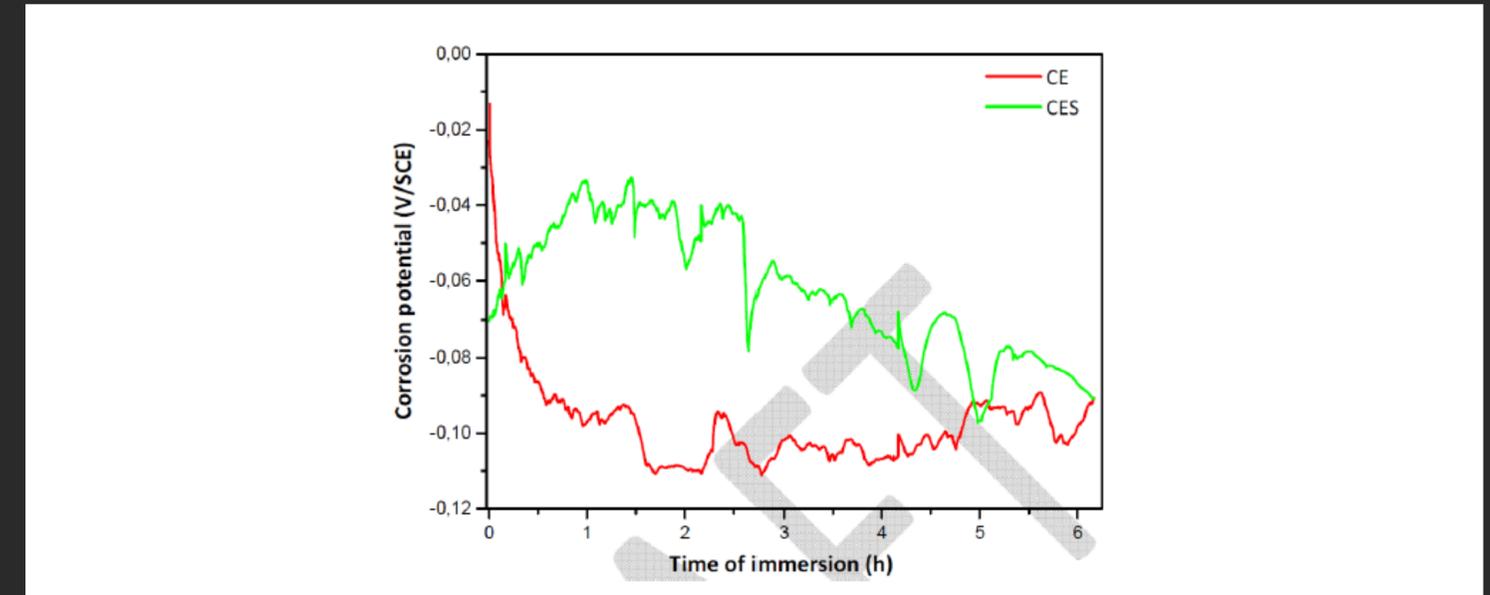
Measure of open circuit potential after 0 2 4 and 6 hours of immersion.

Measure of polarization resistance after 0 2 and 4 hours of immersion.

Measure of anodic polarization after 6 hours of immersion.

## CORROSION TEST RESULTS.

CORROSION POTENTIAL VS IMMERSION TIME



The traditionally EP sample becomes less noble over time (red curve)

The dry EP sample becomes more noble until 2 hours of immersion and then decreases progressively (green curve)

After 6 hours, the dry EP sample and the EP sample have the same corrosion potential ( $E_{\text{corr}}$ )

The dry EP sample has a better protection to corrosion until 6 hours of immersion (green curve above the red curve)

# CORROSION TEST RESULTS.

POLARISATION RESISTANCE VS IMMERSION TIME

Immersion time	0h	2h	4h
CE	0,45 kΩ	2,83 kΩ	0,91 kΩ
CES	7,87 kΩ	12,40 kΩ	14,34 kΩ

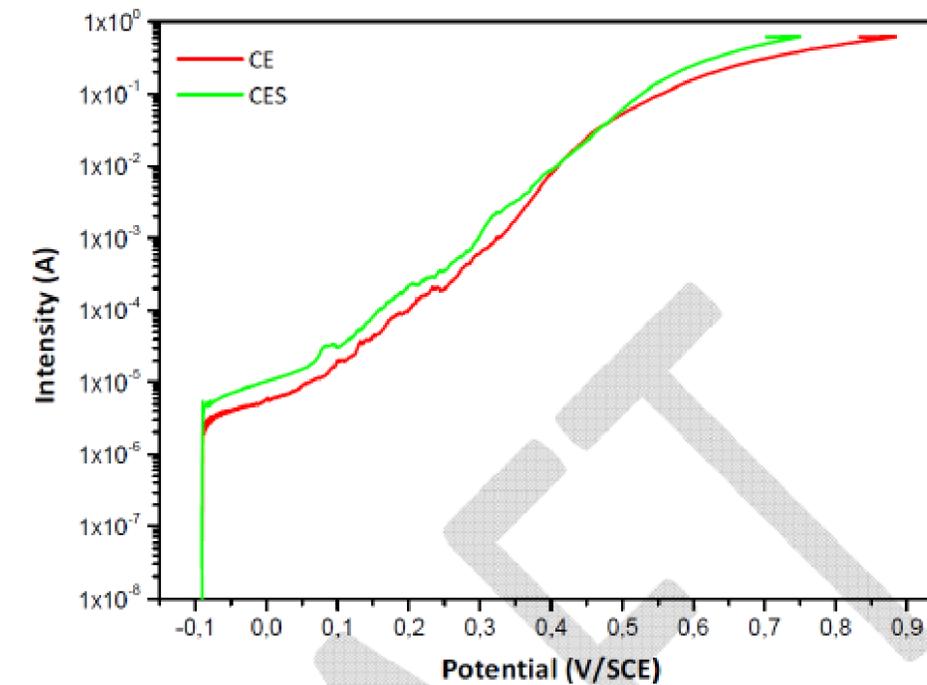
Table 2: Values of polarisation resistance depending on immersion time

Up to 4h of immersion, the dry EP sample has a higher polarisation resistance ( $R_p$ ).

As  $R_p$  is inversely proportionnal to the corrosion rate, the dry EP sample corrodes between 4 to 15 times slower than the traditionally EP sample

# CORROSION TEST RESULTS.

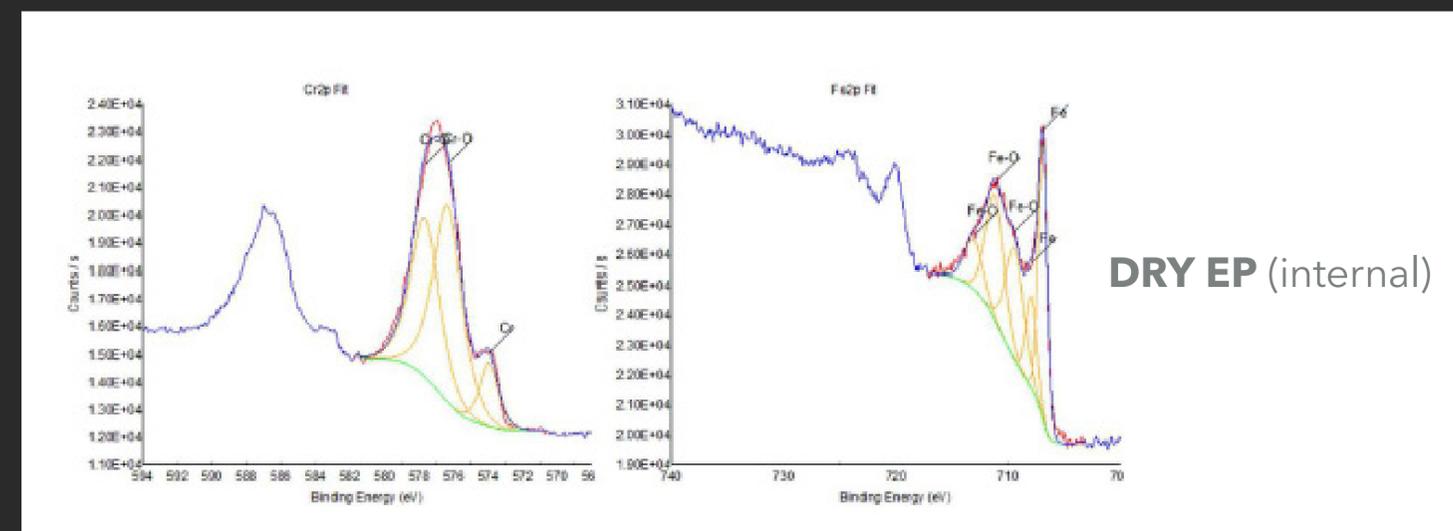
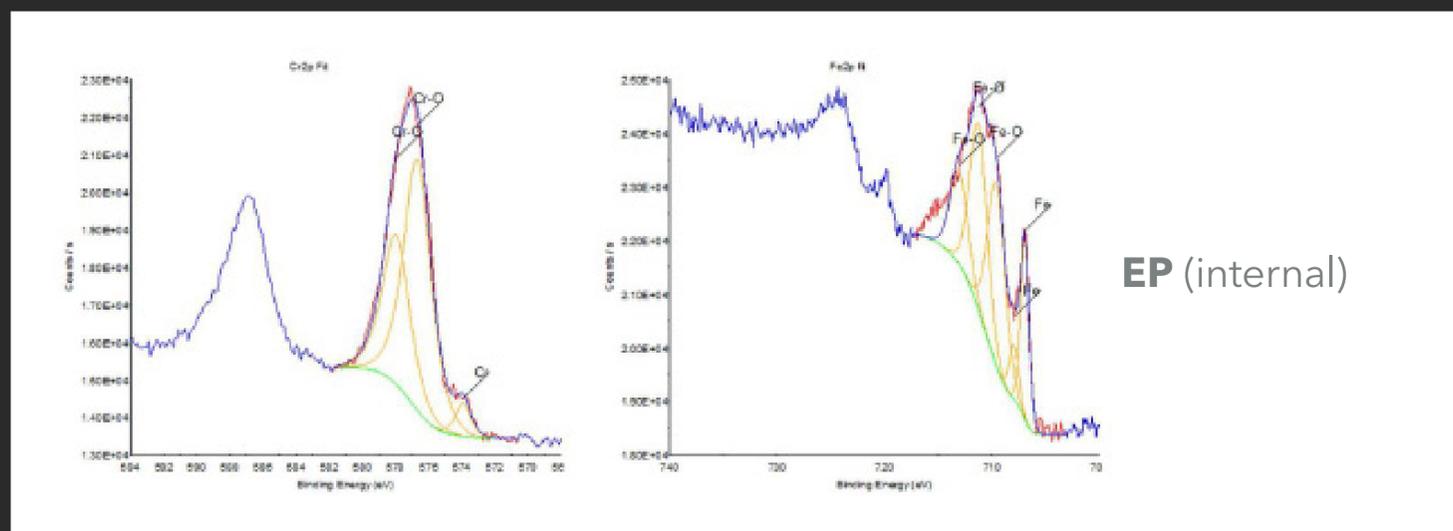
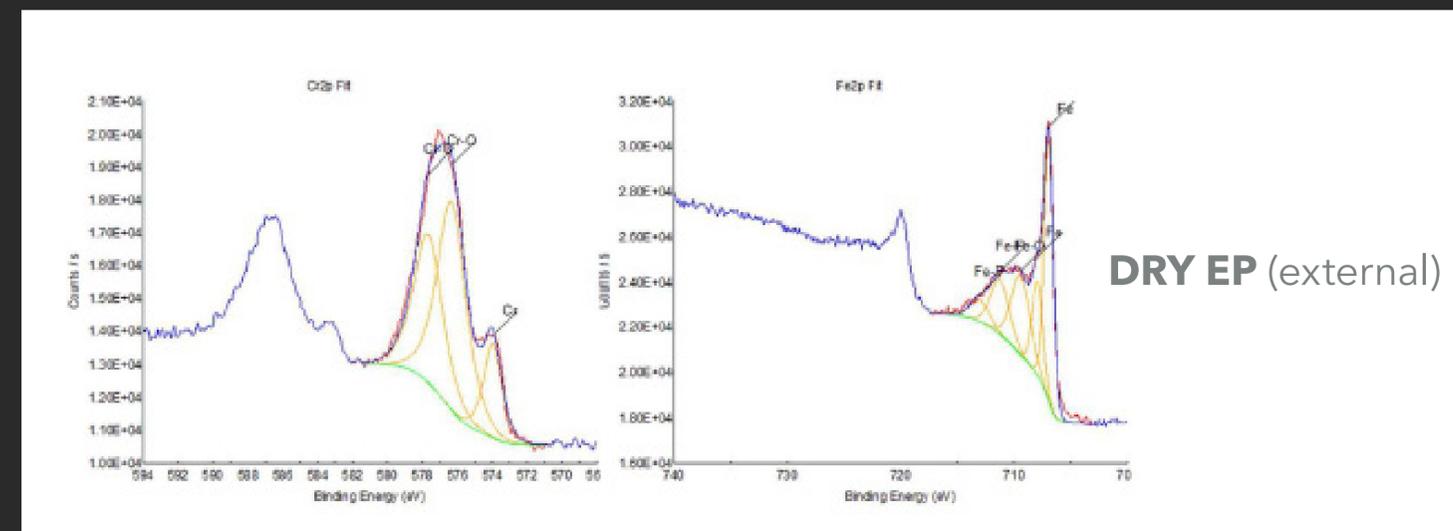
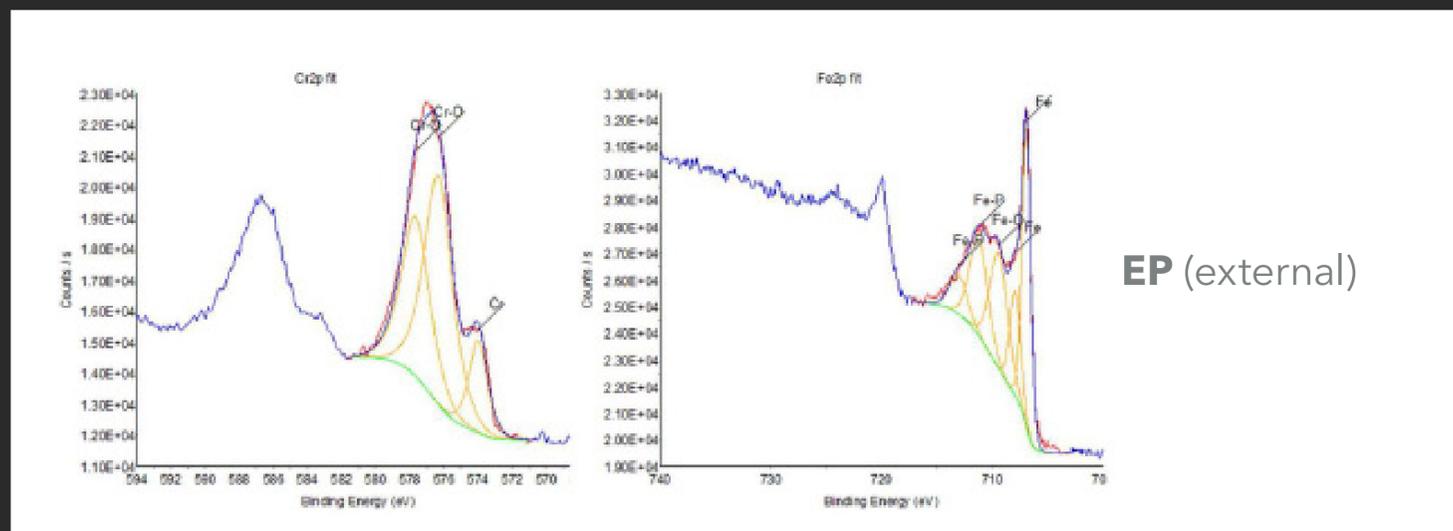
ANODIC POLARISATION AFTER 6 HOURS OF IMMERSION



The two samples show the same behaviour after 6 hours of immersion.

# XPS ANALYSIS RESULTS.

## SURFACE CHEMICAL COMPOSITION



## XPS ANALYSIS RESULTS.

### SURFACE CHEMICAL COMPOSITION

Peak fit carried out according to SEMASPEC #90120403B-STD procedure

	Cr(ox)/Fe(ox)
<b>EP</b> (external)	1.4
<b>EP</b> (internal)	1.3
<b>DRY EP</b> (external)	1.7
<b>DRY EP</b> (internal)	1.5

The oxide component ratio Cr(ox)/Fe(ox) is similar on the EP and DRY EP samples.

The dry EP sample has been successfully electropolished on the external and internal surfaces.

## XPS ANALYSIS RESULTS.

### OXIDE THICKNESS DETERMINATION

	Oxide thickness (nm)	
	Chromium oxide	Iron oxide
<b>EP</b> (external)	5.0	2.0
<b>EP</b> (internal)	7.1	3.6
<b>DRY EP</b> (external)	4.8	1.7
<b>DRY EP</b> (internal)	5.5	2.3

The dry EP sample has been successfully electropolished on the internal and external surfaces.

The chromium oxide thickness is superior to the iron oxide thickness.

## CONCLUSIONS.

The parts treated by dry EP are shinier.

Dry EP affects the external and internal surfaces (no need for internal electrodes): verified by oxide thickness and oxide ratio measurements.

Dry EP gives a better resistance to corrosion up to a certain duration, after which the sample shows the same behaviour than a traditionally EP sample.

Technically, the dry EP process of GPA Innova could be a good alternative to the traditional EP.